



Some Results on Contra Harmonic Index of Some Molecular Graphs

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Abstract: Topological index is a mathematical coding of the molecular graphs that predicts the physicochemical, biological, toxicological and structural properties of chemical compounds that are directly associated with molecular graphs. In this paper, we find the Contra Harmonic index of some graphs which are also the chemical structures of certain alkanes.

Keywords: *Graphs, Contra Harmonic Mean graphs, Contra Harmonic index, Path, Cycle, Triangular chain, Corona product*

1. Introduction

Chemical graph theory is a pivotal branch of mathematical chemistry where molecular structures are modelled as graphs called molecular graphs, with atoms represented as vertices and chemical bonds as edges. A primary tool in this field is the topological index which is a numerical descriptor derived from a graph's structure that correlates with the physicochemical properties of the corresponding molecule. These indices play a crucial role in Quantitative Structure-Property Relationship (QSPR) and Quantitative

Structure-Activity Relationship (QSAR) studies, enabling the prediction of properties like boiling point, stability and biological activity without resorting to expensive lab experiments.

The pioneering Wiener index introduced by Harold Wiener in the year 1947, conceived for modeling alkane boiling points, remains historically significant for its focus on distance sums [8]. Subsequent development emphasized degree-based indices. The Zagreb indices [1] linked to molecular π -electron energy and the highly successful Randić index generated extensive research. Each index provides a unique lens on molecular structure, prompting the exploration of related mathematical means. The Contra Harmonic index is a more recent addition defined via the Contra Harmonic mean of vertex degrees [4]. Here, initially we identify the Contra Harmonic index of fundamental molecular graphs. A simple graph can be moulded into a chemical structure by using some graph operations. So, many chemical graphs can be generated by applying graphical operations on simple graphs, for instance, alkane (C_3H_6) is the corona product of P_3 and $K_{1,2}$.

Definition 1.2: Contra Harmonic index of a graph G is defined as sum of the term $\frac{d(u)^2+d(v)^2}{d(u)+d(v)}$ over all edges uv of graph G .

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$$CH(G) = \sum_{uv \in E(G)} \frac{d(u)^2 + d(v)^2}{d(u) + d(v)}$$

2. Results

Theorem 2.1: Contra Harmonic index of Cycle C_n is $2n$

Proof

Let C_n be a Cycle with n vertices and n edges

Let u_1, u_2, \dots, u_n be vertices of C_n such that u_1 is adjacent to u_2, u_2 is adjacent to u_3, \dots, u_{n-1} is adjacent to u_n and u_n is adjacent to u_1

Then $|E(G)| = n$ and $d(u_i) = 2$, for all i

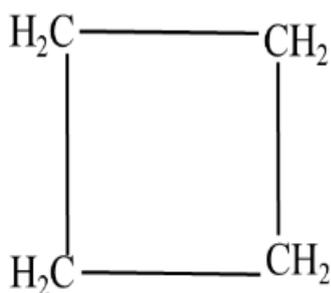
$$\begin{aligned} CH(C_n) &= \sum_{u_i u_j \in E(G)} \frac{d(u_i)^2 + d(u_j)^2}{d(u_i) + d(u_j)} \\ &= \frac{d(u_1)^2 + d(u_2)^2}{d(u_1) + d(u_2)} + \dots + \frac{d(u_{n-1})^2 + d(u_n)^2}{d(u_{n-1}) + d(u_n)} + \frac{d(u_n)^2 + d(u_1)^2}{d(u_n) + d(u_1)} \\ &= \frac{2^2 + 2^2}{2 + 2} + \dots + \frac{2^2 + 2^2}{2 + 2} \quad (n \text{ times}) \\ &= n \left(\frac{2^2 + 2^2}{2 + 2} \right) \\ &= 2n \end{aligned}$$

Therefore, $CH(C_n) = 2n$

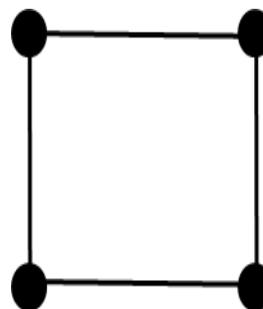
Example 2.1

The following example shows the Contra Harmonic index of cyclobutane using simple

graph (hydrogen depleted graph) C_4 . Here, $CH(C_4)$ is 8



a) Lewis structure of cyclobutane



b) simple graph C_4 (hydrogen depleted graph) of cyclobutane

Figure 1

Theorem 2.2: Contra Harmonic index of Path P_n is $2(n - 3) + \frac{10}{3}, n \geq 3$

Proof

Let P_n be a Path

Let u_1, u_2, \dots, u_n be vertices of P_n such that u_i is adjacent to u_{i+1} , for all $2 \leq i \leq n - 1$

Then $|E(G)| = n - 1$ and $d(u_1) = 1, d(u_n) = 1, d(u_i) = 2$, for all $2 \leq i \leq n - 1$

$$\begin{aligned}
CH(P_n) &= \sum_{u_i u_j \in E(G)} \frac{d(u_i)^2 + d(u_j)^2}{d(u_i) + d(u_j)} \\
&= \sum_{\substack{u_i u_{i+1} \in E(G) \\ i=1, n-1}} \frac{d(u_i)^2 + d(u_{i+1})^2}{d(u_i) + d(u_{i+1})} + \sum_{\substack{u_i u_{i+1} \in E(G) \\ i \neq 1, n-1}} \frac{d(u_i)^2 + d(u_{i+1})^2}{d(u_i) + d(u_{i+1})} \\
&= 2 \left(\frac{1^2 + 2^2}{1 + 2} \right) + (n-3) \left(\frac{2^2 + 2^2}{2 + 2} \right) \\
&= 2 \times \frac{5}{3} + (n-3) \left(\frac{2^2 + 2^2}{2 + 2} \right) \\
&= \frac{10}{3} + (n-3) \left(\frac{2^2 + 2^2}{2 + 2} \right)
\end{aligned}$$

Therefore, $CH(P_n) = 2(n-3) + \frac{10}{3}, n \geq 3$

Example 2.2:

The following example shows the Contra Harmonic index of alkane using simple graph

(hydrogen depleted graph) P_4 . Here, $CH(P_4)$ is $\frac{16}{3}$.



a) Lewis structure of alkane



b) simple graph P_4 of alkane (hydrogen depleted graph)

Figure 2

Theorem 2.3: Contra Harmonic index of $C_n \odot K_{1,2}$ is $\frac{54n}{5}$.

Proof

Let C_n be a cycle with vertices u_1, u_2, \dots, u_n and let v_i, w_i be the pendant vertices adjacent to $u_i, 1 \leq i \leq n$. Then $d(u_i) = 4$ and $d(v_i) = d(w_i) = 1$

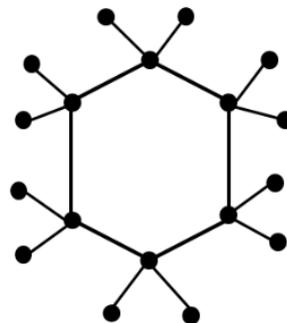
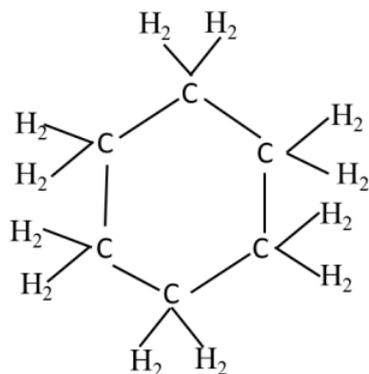
$$\begin{aligned}
CH(C_n \odot K_{1,2}) &= \sum_{uv \in E(G)} \frac{d(u)^2 + d(v)^2}{d(u) + d(v)} \\
&= \sum_{u_i w_i \in E(G)} \frac{d(u_i)^2 + d(w_i)^2}{d(u_i) + d(w_i)} + \sum_{u_i v_i \in E(G)} \frac{d(u_i)^2 + d(v_i)^2}{d(u_i) + d(v_i)} + \sum_{u_i u_j \in E(G)} \frac{d(u_i)^2 + d(u_j)^2}{d(u_i) + d(u_j)} \\
&= n \left(\frac{4^2 + 1^2}{4 + 1} \right) + n \left(\frac{4^2 + 1^2}{4 + 1} \right) + n \left(\frac{4^2 + 4^2}{4 + 4} \right)
\end{aligned}$$

Therefore, $CH(C_n \odot K_{1,2}) = \frac{54n}{5}$

Example 2.3:

The following example shows the Contra Harmonic index of cyclohexane C_6H_{12} using

molecular graph $C_6 \odot K_{1,2}$. Here, $CH(C_6 \odot K_{1,2}) = CH(C_6H_{12})$ is $\frac{324}{5}$.



a) Lewis structure of cyclohexane C_6H_{12}

Molecular graph of cyclohexane C_6H_{12} ($C_6 \odot K_{1,2}$)

Figure 3

Theorem 2.4: Contra Harmonic index of $P_n \odot K_{1,2}$ is

$$\frac{54(n-3)}{5} + \frac{120}{7}, n \geq 3$$

Proof

Let P_n be a path with vertices u_1, u_2, \dots, u_n and let v_i, w_i be the pendant vertices adjacent to u_i , $1 \leq i \leq n$

Then $d(u_1) = d(u_n) = 3$, $d(u_i) = 4$, $2 \leq i \leq n-1$ and $d(v_i) = d(w_i) = 1$

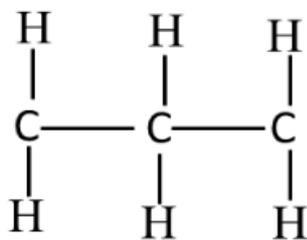
$$\begin{aligned} CH(P_n \odot K_{1,2}) &= \sum_{uv \in E(G)} \frac{d(u)^2 + d(v)^2}{d(u) + d(v)} \\ &= \sum_{\substack{u_i w_i \in E(G) \\ i \neq 1, n}} \frac{d(u_i)^2 + d(w_i)^2}{d(u_i) + d(w_i)} + \sum_{\substack{u_i v_i \in E(G) \\ i \neq 1, n}} \frac{d(u_i)^2 + d(v_i)^2}{d(u_i) + d(v_i)} + \sum_{\substack{u_i w_i \in E(G) \\ i=1, n}} \frac{d(u_i)^2 + d(w_i)^2}{d(u_i) + d(w_i)} \\ &\quad + \sum_{\substack{u_i v_i \in E(G) \\ i=1, n}} \frac{d(u_i)^2 + d(v_i)^2}{d(u_i) + d(v_i)} + \sum_{\substack{u_i u_{i+1} \in E(G) \\ i=1, n-1}} \frac{d(u_i)^2 + d(u_{i+1})^2}{d(u_i) + d(u_{i+1})} \\ &\quad + \sum_{\substack{u_i u_{i+1} \in E(G) \\ 2 \leq i \leq n-2}} \frac{d(u_i)^2 + d(u_{i+1})^2}{d(u_i) + d(u_{i+1})} \\ &= (n-3) \left(\frac{4^2 + 1^2}{4+1} \right) + (n-3) \left(\frac{4^2 + 1^2}{4+1} \right) + 2 \left(\frac{3^2 + 1^2}{3+1} \right) + 2 \left(\frac{3^2 + 1^2}{3+1} \right) + 2 \left(\frac{3^2 + 4^2}{3+4} \right) + (n-3) \left(\frac{4^2 + 4^2}{4+4} \right) \\ &= (n-3) \left(\frac{17}{5} + \frac{17}{5} + 4 \right) + 2 \left(\frac{10}{4} + \frac{10}{4} + \frac{25}{7} \right) \\ &= \frac{54(n-3)}{5} + \frac{120}{7} \end{aligned}$$

Therefore, $CH(P_n \odot K_{1,2}) = \frac{54(n-3)}{5} + \frac{120}{7}, n \geq 3$

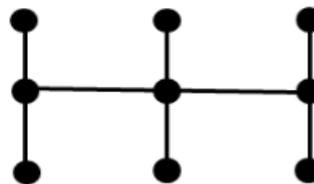
Example 2.4:

The following example shows the Contra Harmonic index of cyclohexane C_6H_{12} using

molecular graph $P_3 \odot K_{1,2}$. Here,
 $CH(P_3 \odot K_{1,2}) = CH(C_3H_6)$ is $\frac{120}{7}$.



a) Lewis structure of alkane C_3H_6



b) Molecular graph of alkane C_3H_6

Figure 4

Theorem 2.5: Contra Harmonic index of $T_1(n)$ is $\frac{32n}{3} - 4$

Proof

Let u_1, u_2, \dots, u_n be vertices in base of the triangles of $T_1(n)$ and let v_1, v_2, \dots, v_{n-1} be the vertices in the peak of the triangles of $T_1(n)$

Then $d(u_1) = d(u_n) = 2$, $d(v_i) = 2$, $1 \leq i \leq n-1$ and $d(u_i) = 4$, $2 \leq i \leq n-1$

$$\begin{aligned} CH(T_1(n)) &= \sum_{uv \in E(G)} \frac{d(u)^2 + d(v)^2}{d(u) + d(v)} \\ &= \sum_{\substack{u_i v_j \in E(G) \\ i=1, n}} \frac{d(u_i)^2 + d(v_j)^2}{d(u_i) + d(v_j)} + \sum_{\substack{u_i v_j \in E(G) \\ i \neq 1, n}} \frac{d(u_i)^2 + d(v_j)^2}{d(u_i) + d(v_j)} + \sum_{\substack{u_i u_j \in E(G) \\ i=1, n}} \frac{d(u_i)^2 + d(u_j)^2}{d(u_i) + d(u_j)} \\ &\quad + \sum_{\substack{u_i u_j \in E(G) \\ i \neq 1, n}} \frac{d(u_i)^2 + d(u_j)^2}{d(u_i) + d(u_j)} \\ &= 2 \times \left(\frac{2^2 + 2^2}{2+2} \right) + 2(n-1) \left(\frac{4^2 + 2^2}{4+2} \right) + 2 \left(\frac{4^2 + 2^2}{4+2} \right) + (n-2) \left(\frac{4^2 + 4^2}{4+4} \right) \\ &= (2 \times 2) + 2n \left(\frac{20}{6} \right) + 4(n-2) \end{aligned}$$

Therefore, $CH(T_1(n)) = \frac{32n}{3} - 4$

Example 2.5:

The following example shows the Contra Harmonic index of $T_1(4)$. Taking several layers

of $T_1(n)$ forms a boron nanotube. Here,
 $CH(T_1(4))$ is $\frac{116}{3}$

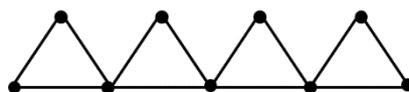


Figure 5: $T_1(4)$

3. Conclusion

In this paper, we present some applications of obtained results for

particular chemical structures such as alkanes and cycloalkanes. Identifying the indices of simple graph structures simplify

the identification of complex chemical compounds.

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